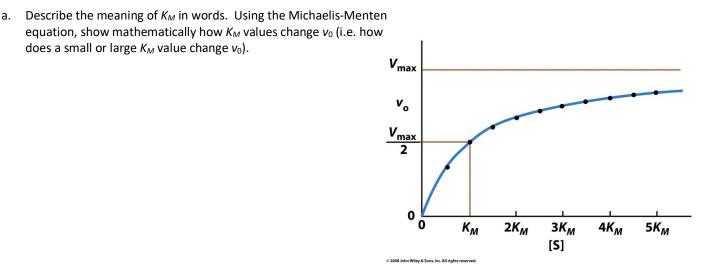
1) (20 points) Consider the following plot of initial velocity of a reaction versus substrate concentration:



b. Using the Michaelis-Menten equation, show mathematically why at low substrate concentrations v_0 is very small.

c. Using the Michaelis-Menten equation, show mathematically why $v_0 = \frac{1}{2} V_{max}$ when $K_M = [S]$

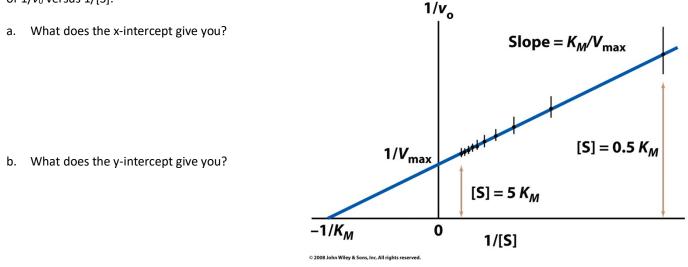
d. Using the Michaelis-Menten equation, show mathematically why at high substrate concentrations, $v_0 \approx V_{max}$.

e. At high [S], k_2 becomes the rate limiting step of the reaction. Knowing this information, mathematically explain how $K_M \approx K_S$.

f. Using the Michaelis-Menten equation, explain why it would be good to have a very small K_M (K_S) value.

Applications of Enzyme Kinetics:

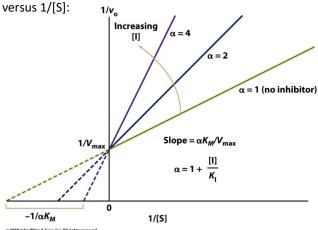
2) (10 points) Consider the following variation of the Michaelis-Menten equation, and the corresponding Lineweaver-Burk plot of 1/v₀ versus 1/[S]:



- c. Why does the data tend to cluster on the left hand side of the graph?
- d. What is the propensity for error on the right side of the plot and what significance does that have?
- e. Why is this type of a plot useful experimentally?

Applications of Enzyme Kinetics:

- 3) **(20 points)** Consider the following Lineweaver-Burk plot of $1/v_0$ versus 1/[S]:
 - a. What type of inhibition is this, and how can you tell?



b. How does this mechanism of inhibition work? Be sure to address where the inhibitor binds, and how this affects the enzyme.

c. Describe:

- in words what the parameter "α" is
- what its limits (i.e its maximum and minimum values) are
- how K_1 can impact the magnitude of α
- what impact α has on the observed Linweaver-Burke plot.
- d. Describe what happens to the following terms with increasing [I]:
 - K_M
 - α
 - V_{max}
- e. What is the capability of increasing [S] to overcome the inhibition?